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Pd-catalyzed *ortho*-arylation of 3,4-dihydroisoquinolones via C–H bond activation: synthesis of 8-aryl-1,2,3,4-tetrahydroisoquinolines

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1. Introduction

The biaryl scaffold has been found in various biologically active compounds¹ and valuable building blocks for synthetic polymers.² Currently, the most widely used methods for the formation of aryl–aryl bonds involve Pd-catalyzed cross-coupling reactions between aryl halides and arylmetals.^{3,4} However, these transformations suffer from the disadvantage that they require installation of functional groups on both coupling partners, which results in expensive and longer synthetic routes to prepare starting materials. In this sense, regioselective direct coupling of aromatic C–H with functionalized arenes has a tremendous potential from the viewpoint of atom economy and efficiency. Recently, transition metal-catalyzed C–H bond activation/arylation methods have been shown to be effective for C_{aryl}–H/C_{aryl}–X cross-coupling reactions.⁵

In our continuous effort to build a privileged scaffold library containing a biaryl moiety, we focused our attention on the family of aporphinoids. Aporphine is a large group of benzylisoquinolinederived alkaloids with more than 500 members, characterized by the tetracyclic skeleton shown in Figure 1. The relevant structural features of these alkaloids are the presence of an isoquinoline core, together with a biaryl subunit and one stereogenic center at C-6a. A wide range of interesting biological activities have been studied, including anticancer,⁶ antimalarial,⁷ antiplatelet,⁸ vasorelaxing,⁹

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ABSTRACT

An efficient route to synthesize biologically interesting 8-aryl-1,2,3,4-tetrahydroisoquinoline has been developed. It involves the Pd-catalyzed direct arylation of 3,4-dihydroisoquinolones via C–H bond activation with aryl iodides to afford a variety of 8-arylated cross-coupling products, which are subsequently reduced to 8-aryl-1,2,3,4-tetrahydroisoquinolines in good to excellent yields.

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and leischmanicidal activities.¹⁰ These various pharmacological activities have prompted intensive structure–activity relationship studies over the past three decades.¹¹

Despite many methods for the synthesis of isoquinolines¹² and tetrahydroisoquinolines,¹³ only one attempt was made by Ellefson using aryloxazolines¹⁴ as key intermediates to access the 8-aryl-1,2,3,4-tetrahydroisoquinolines, which possess the basic structural features of the aporphine skeleton without the C-7 methylene unit.¹⁵ This limited synthetic route prompted us to investigate more efficient and convergent approaches to 8-aryl-1,2,3,4-tetrahydroisoquinolines. We now report our success.

2. Results and discussion

The key transformations in the synthesis of 8-aryl-1,2,3,4-tetrahydroisoquinolines involve the introduction of an aryl group at the



Figure 1. Aporphine alkaloids and 8-aryl-1,2,3,4-tetrahydroisoquinoline.



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a. Pd-catalyzed *ortho*-arylation *via* C-H bond activation; **b**. reduction

Scheme 1. Key transformations.



Scheme 2. Synthesis of 3,4-dihydroisoquinolones.

C-8 position regioselectively directed by a carbonyl group via C–H bond activation and subsequent reduction (Scheme 1). Recently, many research groups have shown a significant progress in the regioselective C–H bond activation of arenes containing a directing group under Pd, Ru, or Rh catalysis.¹⁶ However, the direct arylation of 3,4-dihydroquinolones is unprecedented to the best of our knowledge.

The 3,4-dihydroisoquinolones (**1–6**) required for this methodology are readily prepared by a modified protocol shown in Scheme 2.

With various 3,4-dihydroisoquinolones in hand, the direct arylation of 3,4-dihydroisoquinolones with aryl iodides was investigated. The initial optimization was carried out with respect to 1 with 4-iodoacetophenone using Pd catalysis in different solvents. To our delight, the regioselective arylation went quite smoothly using Pd(OAc)₂ as a catalyst and AgOAc as an additive at the elevated temperature to give 8-arylated adduct 8 in 83% isolated vield. The regioselectivity was confirmed with other known compounds in the literature.¹⁷ The optimized conditions for the regioselective direct arylation are the combination of 1.0 equiv of substrate, 3.0 equiv of aryl iodide, 1.3 equiv of AgOAc as an additive, 5 mol % of Pd(OAc)₂, and trifluoroacetic acid as a solvent. The results are summarized in Table 1.¹⁸ The reactions proceeded well at the elevated temperatures (110-130 °C) in 5-24 h, depending on the substitution patterns in the 3,4-dihydroisoquinolones and the functional groups on the aryl iodides. Regarding the 3,4dihydroisoquinolones, electron-rich substrates 1 and 4 reacted faster than unsubstituted or more hindered substrates **2** and **3** (Table 1, compare entries 8, 13, and 18). Electron-poor substrate **5** failed to give any desired products. This implies that the electron-nature of the aromatic ring is also a critical factor for the successful C–H bond activation. In the case of *N*–H substrate **6**, most of the starting material was recovered intact, and only trace amounts of cross-coupling product were isolated (<10%). Presumably, the nitrogen atom in the amide coordinates with the palladium catalyst, resulting in much lower yields. Good results have been obtained with both electron-rich and electron-poor aryl iodides (entries 2–5). It is noteworthy that 4-bromo-iodobenzene also proceeds smoothly to afford bromo-substituted product **12**, which could be further modified by Heck¹⁹ and Suzuki³ reactions (entry 6).

After we have proven the efficiency of this process to access a variety of 8-aryl-3,4-dihydroisoquinolones, we next reduced these cross-coupling products into the corresponding 1,2,3,4-tetrahydro-isoquinolines with LiAlH₄ in THF for 2 h. The selected results are shown in Table 2.²⁰

3. Conclusions

In conclusion, we have developed a new, concise, and convergent procedure for biologically interesting 8-aryl-1,2,3,4-tetrahydroisoquinolines from readily available starting materials. Extensions of this method to the synthesis of other biologically interesting biaryls are underway.

Table 1

Synthesis of 8-aryl-3,4-dihydroisoquinolones by Pd-catalyzed ortho-arylation^a



Table 1 (continued)

Entry	Substrate	Aryl iodide	Temp time	Product	Yield ^b (%)
7	2	I — Ac	110 °C 24 h	$ \begin{array}{c} $	55
8	2	I — CI	110 °C 17 h		73
9	2	I - OCF3	110 °C 17 h	0 15 OCF ₃	63
10	2	I-CF3	110 °C 14 h	0 0 16 CF ₃	70
11	3	I — Ac	120 °C 15 h	0 0 0 0 17 Ac	50
12	3	I-CO2Et	120 °C 15 h	$ \begin{array}{c} $	59

Table 1 (continued)

Entry	Substrate	Aryl iodide	Temp time	Product	Yield ^b (%)
13	3	I————————————————————————————————————	110 ℃ 20 h		63
14	3	I-CF3	110 °C 16 h	0 0 0 0 20 CF ₃	62
15	4	I — Ac	110 °C 7 h	0 0 21 Ac	50
16	4	I-CO2Et	110 °C 5 h	→O → N → O → 22 CO ₂ Et	68
17	4	I-OCF3	110 °C 7 h	O V V V V V V V V V V V V V	80
18	4	I-CI	110 °C 7 h		76

^a All reactions were run under the following conditions unless otherwise specified: substituted 3,4-dihydroisoquinolone (1.0 equiv), aryl iodide (3.0 equiv), 5 mol % Pd(OAc)₂, AgOAc (1.3 equiv), trifluoroacetic acid (1.25 mL/substrate mmol). ^b Isolated yields after column chromatography.

Table 2

Reduction to 8-aryl-1,2,3,4-tetrahydroisoquinolines



^a Isolated yields after column chromatography.

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¹H NMR (400 MHz, CDCl₃) δ 7.40–7.38 (m, 2H), 7.31–7.29 (m, 3H), 6.63 (s, 1H), 5.94 (s, 2H), 3.53 (t, *J* = 6.4 Hz, 2H), 3.02 (s, 3H), 2.92 (t, *J* = 6.4 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 163.7, 149.0, 145.5, 136.3, 135.3, 128.8, 127.8, 127.3, 125.5, 121.9, 106.3, 101.5, 48.0, 35.2, 29.7; IR (film) 2896, 1648, 1458, 1259, 1054 cm⁻¹; TLC *R*_f (hexanes:EtOAc 1:1) = 0.44; HRMS (ESI) *m/z* calcd for C₁₇H₁₆NO₃ (M+H)^{*} 282.1130, found 282.1134.

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- General Procedure for reduction of 8-aryl-3,4-dihydroisoquinolones to 8-aryl-1,2,3,4-tetrahydroisoquinolines: A solution of 8-aryl-3,4-dihydroisoquinolone 7

(40 mg, 0.142 mmol) and LiAlH₄ (12 mg, 0.284 mmol) in THF (1.6 mL) was refluxed for 2 h. The reaction mixture was quenched by the successive addition of H₂O (200 µL), 1 N NaOH (200 µL), and H₂O (600 µL) at 0 °C. The mixture was then extracted with CHCl₃ (3 × 10 mL)/H₂O (5 mL). The combined organic layers were washed with brine, and were dried over Na₂SO₄. After filtration and concentration *in vacuo*, the residue was purified via flash column chromatography to compound **25** (33 mg, 87%) as a yellow solid; mp = 82-84 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.44–7.40 (m, 2H), 7.37–7.33 (m, 1H), 7.31–7.29 (m, 2H), 6.59 (s, 1H), 5.84 (s, 2H), 3.26 (s, 2H), 2.90 (t, *J* = 6.4 Hz, 2H), 2.63 (t, *J* = 6.4 Hz, 2H), 2.32 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 145.7, 143.4, 134.3, 129.5, 128.4, 127.7, 127.2, 125.4, 121.5, 107.7, 100.6, 56.5, 52.7, 46.1, 29.7; IR (film) 2931, 2777, 1462, 1375, 1043 cm⁻¹; TLC *R*_f (CH₂Cl₂:MeOH 30:1) = 0.11; RMS (ESI) *m/z* calcd for Cl₁H₁₈NO₂ (M+H)^{*} 268.1338, found 268.1342.